			AQ-699A
В.	Sc. I	art-	-II (Semester-IV) Examination
	4S:	PET	TROCHEMICAL SCIENCE
e—Th	ree I	Hours	[Maximum Marks—80
Note	:	(1)	Question No. 1 is compulsory.
		(2)	All SIX questions carry equal marks.
		(3)	Diagrams and chemical equation should be given wherever necessary.
		(4)	Discuss the reaction mechanism wherever necessary.
(A)	Fill	in th	e blanks :
	(i)	Ethy oxid	vlene glycol is obtained by of ethylene de.
	(ii)	oxio	anese Corp. USA developed process of lation of to obtain ancitic anhydride uired for VAM.
	(iii)		is used as a monomer for acrylic and dacrylic fibers, various resins and rubber.
	(iv)	Pho	as a main by-product. 2

13. (P) Describe Lummus ammoxidation process for manufacture of Terepthalic acid with their chemistry.

(Q) Describe market of following in detail:

Phthalic Anhydride

Terepthalic Acid.

UBS---50071

Time—Three Hours]

(Contd.)

(B)	Cho	ose correct alternative :		9.	. · (P)	Adipic acid is an important chemical in polymer industries. Describe production of adipic acid through cyclohexane.	
	(i)	Direct hydration process for isopropyl alcohol was developed					
		(a) BASF Germany (b) RC			(Q)	 Discuss dehydrogenation of t-Amylenes for production of isoprene with the catalyst, chemistry and process 	
		(c) ICI (UK) (d) Not	ne of these			parameters. 6	
(ii)		Three steps involved in isopre through Goodyear S.D. proces	_	1	0. (A)	 Describe aniline production through ammonolysis process with the process parameters and chemistry involved. 	
(iii)	(1) Dimerization			(B)	In which route acetone has been produced as		
		(2) Isomerization	·			by-product while manufacture of Phenol? Explai	
	(3)				this route in detail with the chemistry.		
		(a) Pyrolysis (b) Hyd	Hydration			OR	
	(c) Dehydrogenation (d) Poly	•	1	1. (P)	Which is main feed-stock for production of Caprolactum? Explain Du-Pont process with the		
	Catalyst used in Acetaldehyde V	Wacker Process			chemistry and process parameters.		
		is:			(Q	Describe liquid phase and vapour phase nitration	
(iv)		(a) CuCl ₂ and V ₂ O ₅ (b) Cu				process for production of aniline with the chemistry and process parameters involved.	
		(c) CuCl ₂ and IO ₂ (d) Cu	Cl ₂ and H ₂ SO ₄		l2. (A) Describe the process of production of phthalic	
	(iv)	Aniline may be produced by the Phenol.	he of		·	anhydride through O-Xylene with their advantages	
		(a) Ammionolysis (b) Hy	dration		(B	Explain chemistry and process parameter fo	
		(c) Dehydrogenation (d) Sul	Sulphonation 2			production of Di-methyl Terepthalete by using	
		(-)	1	ì		P-Xylene.	
						OR	
UBS500)71	2	(Contd.)	; .	UBS—50	0071 5 (Contd.	

www.sgbauonline.com

(C)	Ans	wer in one sentence.	
	(i)	Why propylene cannot be easily oxidize propylene oxide?	ed to
	(ii)	Name any three routes for production isoprene.	on of
	(iii)	What are the applications of Acetaldeh	yde?
	(iv)	What are the disadvantages of hydroperoxic process for production of propylene through cumene?	
!. (A)	Exp	lain oxichlorination process with their advancess parameter, chemistry and catalyst inv	ntages, olved. 6
(B)		scribe production of acetaldehyde by using Cl_2 and $PdCl_2$ in detail.	atalyst 6
		OR	
3. (P)	of '	nat are the various raw material utilised in provingly acctate monomer? Explain Celanese In the chemistry involved.	duction Process 6
(Q)		scribe Union Carbide Process for productively alcohol with its chemistry.	tion of
4. (A)	wit	mpare direct oxidation and chlorohydrin that the advantages and disadvantages for pro- ethylene oxide.	
(B)		plain hydrolysis of ethylene oxide with the ched process parameters.	nemistry 6
		OR	
UBSS0	0071	3	(Contd.)

3

UBS--50071

www.sgbauonline.com

5.	(P)	Discuss chemistry and process parameter of chlorohydrin, process for production of ethylene oxide. Also state uses of ethylene oxide.					
	(Q)	Describe uses, chemistry and process parameter for production of ethnol amines through ethylene oxide. 6					
6.	(Λ)	Explain hydroperoxidation process for production of propylene oxide with the chemistry and process parameters. 6					
	(B)	Describe acrylamide manufacture through hydration process involving chemistry and process parameters.					
	OR						
7.	(P)	Describe acetone manufacture through isopropyl, alcohol including their chemistry, process parameters and uses.					
	(Q)	Describe catalytic hydration process for production of isopropyl alcohol with the chemistry, process parameters and uses.					
8.	(A)	Describe process for manufacture of chloroprene through butadiene with the chemistry and process parameters. Why is its gaining importance? 6					
	(B)	Describe and explain oligomerization process for production of isoprene with the chemistry and process parameters.					
	OR						
UBS	- 500	71 4 (Contd.)					