B.Sc. (Part—II) Semester—IV Examination 4S: PETROCHEMICAL SCIENCE

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Time:	: Thr	ee I	lours	5]			[Maximum Marks: 80			
N	lote	:	(1)	Question No. 1 is comp	ulsory.					
			(2)	Diagram and chemical e	equations sho	ould b	be given wherever necessary.			
			(3)	Discuss the reaction me	chanism who	erevei	r necessary.			
1. (2	A) F	illi	n the	blanks:						
	(i	i)		vlene oxidation is done lene oxide.	in the presen	nce o	f catalyst for the synthesis of			
	(1	ii)	O-x	ylene undergoes	reaction to	form	phthalic anhydride.			
	(1	iii)	Chlo	oroprene is also known a	as					
	(iv)	Adip	oic acid is commercially in	nportant as th	e key	chemical for the production of			
							½×4=2			
(I	B) (lho	noose the correct alternative :							
	(1	i)	2-m	ethyl butadiene is also ki	nown as:					
			(a)	Isoprene		(b)	Chloroprene			
			(c)	Styrene		(d)	Cresol			
	(1	ii)	Cata	alytic dehydrogenation of	isopropyl al	coho	l gives			
			(a)	Ethanol		(b)	Acetone			
			(c)	Acetic acid		(d)	Formaldehyde			
	(1	iii)	Raw	te is:						
			(a)	Acetone		(b)	Vinyl chloride monomer			
			(c)	Ethylene		(d)	Propylene			
	(i	iv)	Acry	nd						
			(a)	Plastic		(b)	Modacrylic fibers			
			(c)	Acrylic acid		(d)	Propylene $\frac{1}{2} \times 4 = 2$			
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	(C)	Answer in one sentence each:
		(i) Which is the important route for the synthesis of Isoprene?
		(ii) Which byproduct is obtained during the manufacture of phenol through cumene?
		(iii) Which is the starting material for the synthesis of phenol by Dow process?
		(iv) Name the catalyst used for the synthesis of adipic acid by using butadiene. 1×4=4
9	(Δ)	Why integrated route gets importance for production of vinyl chloride monomer? Describe with respect to their chemistry and process parameters involved.
	(B)	Which feed stocks are used in production of vinyl acetate monomer? Describe acetylene route in detail.
		OR
- 1	(P)	Describe role of PdCl ₂ and CuCl ₂ in Wacker process in manufacture of acetaldehyde.
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	(Q)	Which is application of Acetaldehyde, Vinyl Chloride and Vinyl Acetate?
4	(A)	Compare chlorohydrin process and direct oxidation process for production of ethylene oxide.
	(B)	Describe hydrolysis process of ethylene oxide with respect to their chemistry and process parameters for production of ethylene glycol,
		OR
e	(Ts)	
5 ,	(P)	Ethanol amines used in detergent and solvent industry. How will we prepare ethanolamine? Describe with respect to their chemistry.
	(Q)	Describe oxidation process for manufacture of ethylene oxide with their chemistry and process parameters.
6	(A)	Describe Hoechst-Uhde process for production of acetone with respect to their chemistry, process parameters and uses.
	(B)	Why is indirect oxidation method employed for production of Propylene Oxide? Explain this process in detail.
		OR
7.	(P)	Describe indirect hydration of propylene involving chemistry, process parameters and uses
	. ,	for production of isopropyl alcohol.
	(Q)	Acrylonitrile is used as monomer of various resins fibers and rubbers. Describe ammoxidation process for their manufacture in detail.

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8.	(A)	Describe two step process for production of chloroprene by using feed stock butadiene with process parameter and chemistry involved.
	(B)	Isoprene is the monomer of natural rubber. Describe synthesis of isoprene developed by Goodyear – Scientific design process in detail.
		OR
9.	(P)	Adipic acid is key chemical for the production of nylons. Explain adipic acid production through cyclohexane through two step processes in detail.
	(Q)	Describe dehydrogenation of t-amylenes for production of isoprene monomer with respect to their chemistry, process parameter and their applications.
10.	(A)	Describe cumene route for production of phenol in detail.
	(B)	Describe nitration process for production of aniline with respect to their chemistry and process parameters.
		OR
11.	(P)	Which feed stocks are used for production of caprolactam? Explain conventional process for production of caprolactam in detail.
	(Q)	Describe chlorination process for production of phenol with their chemistry, process parameters and their uses.
12.	(A)	Describe production of Phthalic anhydride by using O-xylene as raw material with respect to their chemistry, process parameter and advantages.
	(B)	Compare dimethyl terephthalate and terephthalic acid route for production of monomer of polyethylene terephthalate.
		OR
13.	(P)	Describe P-xylene route for production of terephthalic acid in detail.
	(Q)	The process for manufacture of DMT from P-xylene by oxidation and esterification is divided into four steps. Name these steps and discuss in detail.

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