B.Sc. (Part—III) Semester—VI Examination PETROCHEMICAL SCIENCE

Time : Three	Hours]	[Maximum Marks : 80
N.B. :— (1) (2) (3)	Question No. 1 is compulsory. Remaining SIX questions carry 12 marks each Give chemical equations and draw diagrams who	
(i) (ii) (iii)	when visible light is passed through a is unit of wave length. HPLC stands for High Liquid Chroma	
	Main constituent in Natural gas is sose the correct alternative :— Which of following solvents is used in UV-Spec (a) Ethanol (b) Water (c) Benzene (d) All above that is AQI?	
, ,	(a) Air Quality Index (b) Air Quality Index (c) Air Quality Intensity (d) All above In polymerization process catalyst is termed as (a) Promoter (b) Initiator (c) Ion (d) Radical	/e
(iv) VTM—13430	In shift converter all CO is converted into (a) CO ₂ (b) H ₂ O (c) CO ₃ (d) CH ₄	 (Contd.)

(C)	Ans	swer the following question in ONE sentence each :—	
	(i)	Which major catalyst was used in IPCL plant for reforming and isomeriz	ation of Xylene?
	(ii)	Define air pollution.	
	(iii)	What do you mean by synthetic fuels?	
	(iv)	What is homogeneous catalyst?	4
(A)	Des	cribe the following with their units and formula:	
	(a)	Wave length	3
	(b)	Wave Number	3
	(c)	Frequency	3
	(d)	Energy.	3
		OR	
(P)	Disc	cuss the principle of IR spectroscopy.	6
(Q)	Exp	lain the important characteristics of electromagnetic radiation.	6
(A)	Wh	ich are important features of NMR spectroscopy?	6
(B)	Wha	at are the applications of mass spectroscopy?	6
		OR	
(P)	Disc	cuss theory; working and principle of mass spectrometer.	12
(A)	Exp	lain theory of Gas Chromatography.	6
(B)	Hov	w we will compare HPLC and ELC ?	6
		OR	
(P)	Wha	at is Chromatography? Classify with tree diagram.	6
(Q)	Disc	cuss the characteristic features of HPLC.	6
Desc	cribe	the following:—	
(i)	Crac	cking Catalyst	6
(ii)	Hon	nogeneous Catalysts.	6
		OR	
1—134	30	2	(Contd.)
	(A) (P) (Q) (A) (B) (P) (Q) (B) (P) (Q) (i) (ii)	(i) (ii) (iii) (iv) (A) Des (a) (b) (c) (d) (P) Disc (Q) Exp (A) Wh (B) Wh (P) Disc (A) Exp (B) Hov (P) Wh (Q) Disc Describe (i) Crac	 (ii) Define air pollution. (iii) What do you mean by synthetic fuels? (iv) What is homogeneous catalyst? (A) Describe the following with their units and formula: (a) Wave length (b) Wave Number (c) Frequency (d) Energy. OR (P) Discuss the principle of IR spectroscopy. (Q) Explain the important characteristics of electromagnetic radiation. (A) Which are important features of NMR spectroscopy? (B) What are the applications of mass spectroscopy? OR (P) Discuss theory; working and principle of mass spectrometer. (A) Explain theory of Gas Chromatography. (B) How we will compare HPLC and ELC? OR (P) What is Chromatography? Classify with tree diagram. (Q) Discuss the characteristic features of HPLC. Describe the following:— (i) Cracking Catalyst (ii) Homogeneous Catalysts. OR

9.	(P)	What is Catalyst? Describe importance of catalyst in petrochemical industries nowada	ys.
			6
	(Q)	Which catalyst is used in various units for manufacture of synthesis gas?	6
10.	(A)	Explain concept of chemical refinery.	6
	(B)	Describe and explain, natural gas is important feed stock for petrochemical industries.	6
		OR	
11.	(P)	In future hydrogen is considered as an important fuel.	6
	(Q)	"Ecology is contributing to precipitate the energy crisis"? How? Explain in detail.	6
12.	(A)	What is pollution? Describe various types of pollution in detail.	12
		OR	
13.	(P)	Explain Ethanolamine sweetening process for removal of sulphur and its derivaties with	its
		process flow.	12