# M.Sc. (Part-II) Semester-III (CBCS) Examination

#### PHARMACEUTICAL CHEMISTRY

### (Modern Pharm. Analytical Tech.–I)

#### Paper-3SA1

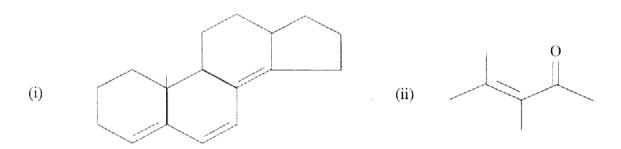
Time : Three Hours]

[Maximum Marks: 80

- Note: (1) ALL questions are compulsory
  - (2) All questions carry equal marks.
  - (3) Draw diagrams wherever necessary.
- (a) With a neat labelled diagram explain design and working principle of double beam UV spectrophotometer.
  - (b) Explain the terms Chromophore and Auxochromes with example. Give the applications of UV spectroscopy.

OR

(p) Using Woodward–Fischer's rule, calculate wavelengths of maximum UV absorption for following compounds:



- (q) Explain Beer-Lambert's law and its deviations.
- 2. (a) Explain the applications of Flame emission spectroscopy in pharmacy. 8
  - (b) Explain instrumentation and applications of AAS.

OR

- (p) Explain the relationship between:
  - (i) Chemical structure and Fluorescent intensity.

8

- (ii) pH and Fluorescent intensity.
- (q) Define Quenching. Write the types of Quenching. Explain relationship between concentration and
  Fluorescent intensity.
- (a) Explain with example how inductive effect, mesomeric effect, resonance and ring strain affect absorption of IR radiation.
  - (b) Explain the various sampling techniques in IR spectroscopy.

8

OR

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(p)	Explain various applications of IR spectroscopy.	8
(q)	Discuss about ATR in detail.	8
(a)	Define chemical shift. Explain shielding and deshielding of protons.	8
(b)	Answer the following	
	(i) Why C12, O16, O18, S32 do not show NMR spectra.	8
	(ii) n-butane shows splitting signals in its NMR spectrum.	
	OR	
(p)	Discuss the following:	
	(i) Natural abundance	8
	(ii) COSY	
(q)	Explain the instrumentation and applications of NMR spectroscopy.	8
(a)	Explain the instrumentation and applications of Mass spectroscopy.	8
(b)	Explain about Ge-Ms in detail.	8
	OR	
(p)	Discuss about chemical ionization mass spectroscopy.	8
(q)	Discuss about fragmentation process and fragmentation pattern in Mass Spectroscopy.	8
	(q) (a) (b) (p) (q) (d) (p) (p)	<ul> <li>(q) Discuss about ATR in detail.</li> <li>(a) Define chemical shift. Explain shielding and deshielding of protons.</li> <li>(b) Answer the following <ul> <li>(i) Why C12, O16, O18, S32 do not show NMR spectra.</li> <li>(ii) n-butane shows splitting signals in its NMR spectrum.</li> </ul> </li> <li>(p) Discuss the following: <ul> <li>(i) Natural abundance</li> <li>(ii) COSY</li> </ul> </li> <li>(q) Explain the instrumentation and applications of NMR spectroscopy.</li> <li>(a) Explain the instrumentation and applications of Mass spectroscopy.</li> <li>(b) Explain about Ge-Ms in detail.</li> </ul> <li>OR</li> <li>(p) Discuss about chemical ionization mass spectroscopy.</li>